

# **INGENIERÍA EN ALIMENTOS**

**GRUPO: 106**

**QUÍMICA GENERAL**



HORNO DE SECADO



PRECAUCION  
SUPERFICIE CALIENTE

RUTA DE EVACUACION

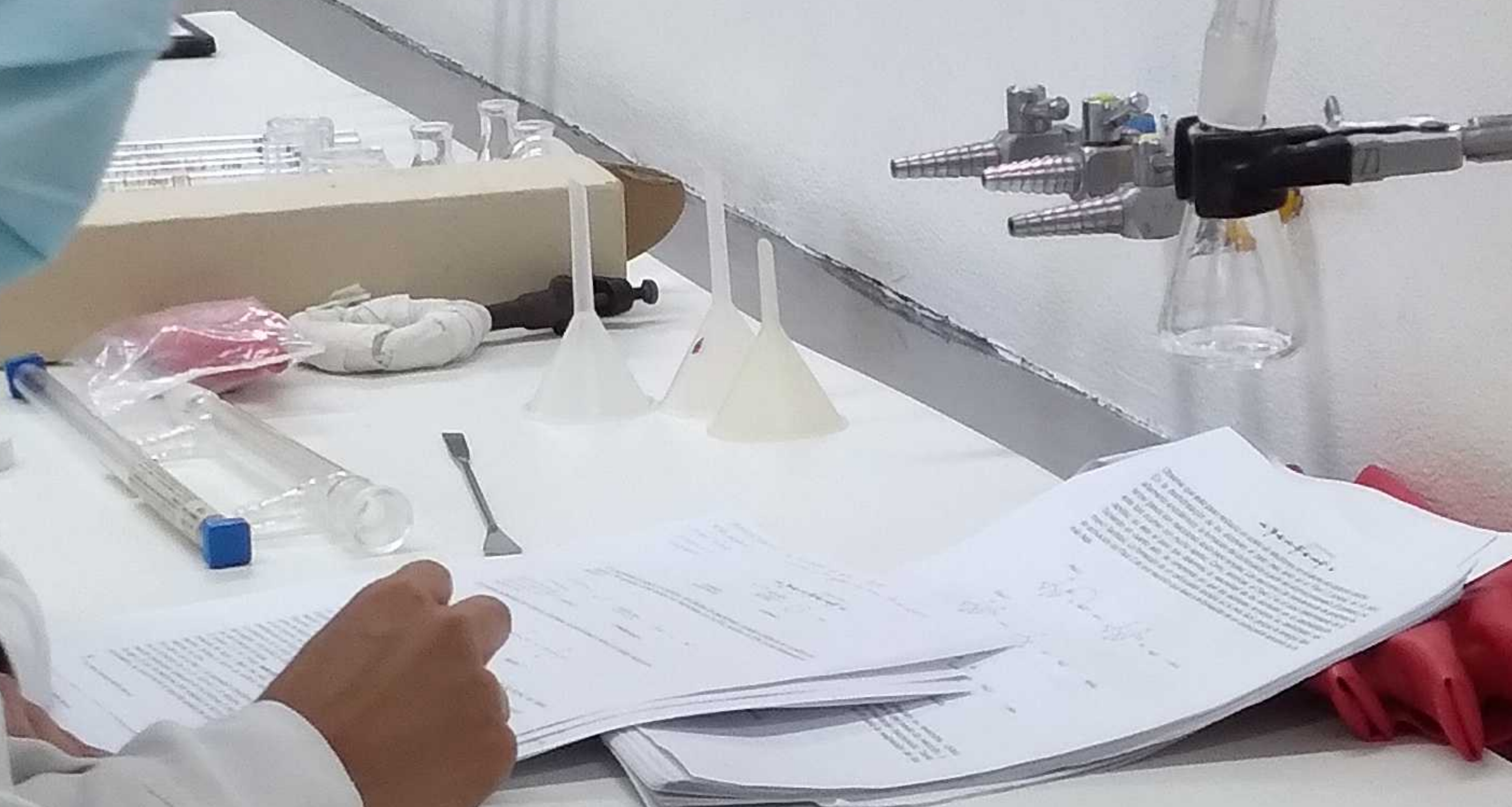
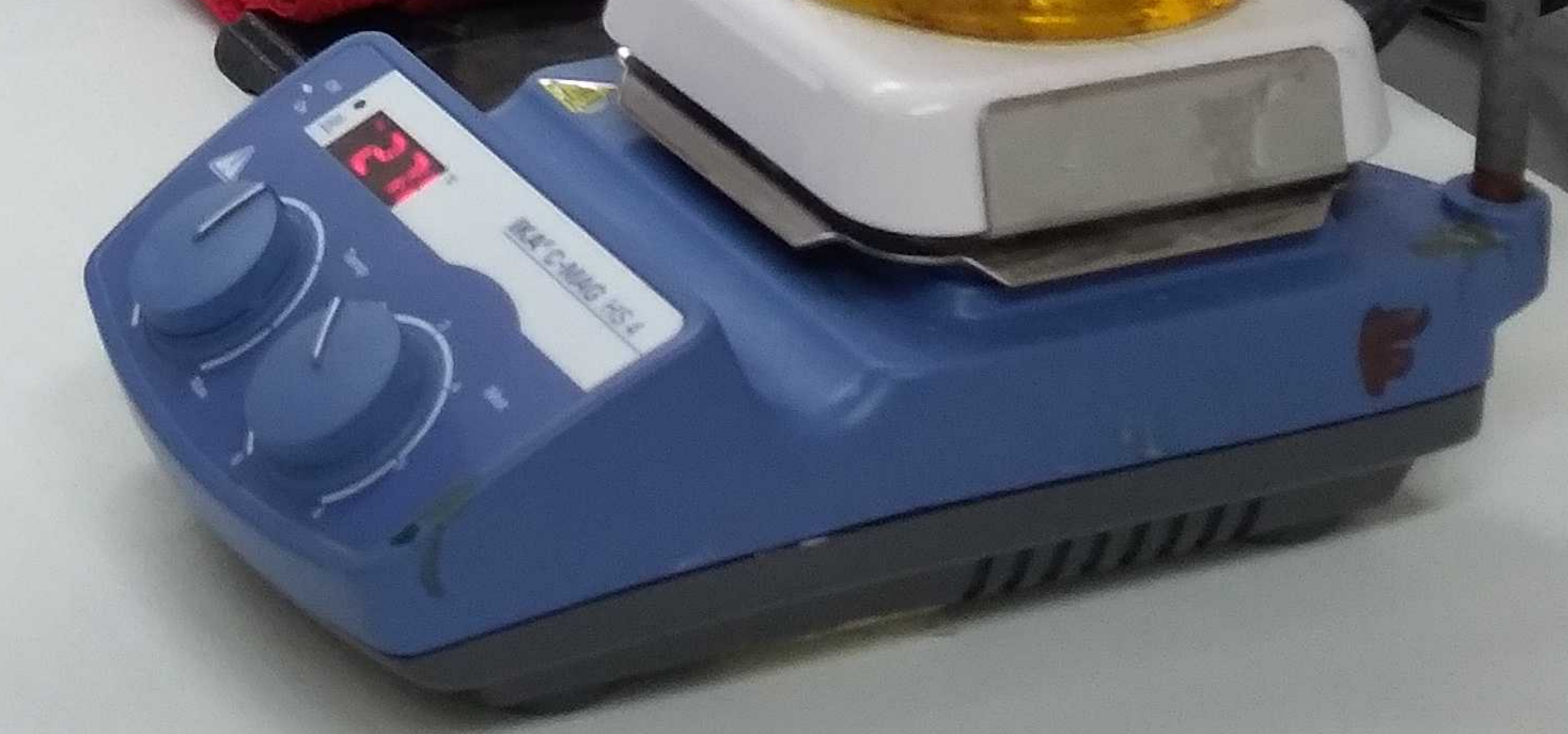
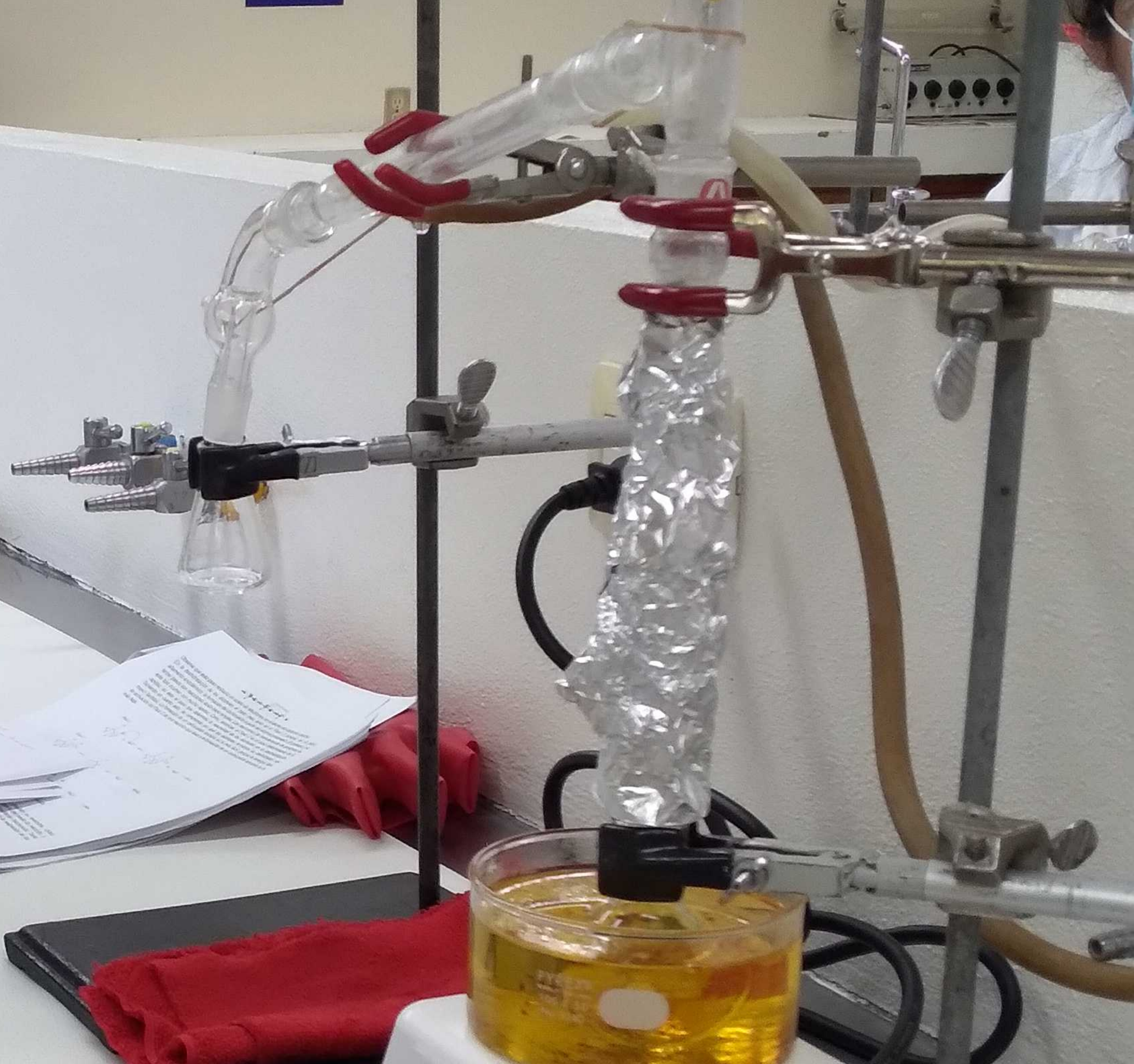
MUFLA

AL FINALIZAR SU TUBO DESCONECTE LA MANGUERA

MATERIAL

USO OBLIGATORIO DE EQUIPO DE PROTECCION PERSONAL

DIGESTOR KJELDAHL





PRECAUCION SUPERFICIE CALIENTE

HORNO DE SECADO

PRECAUCION SUPERFICIE CALIENTE

RUTA DE EVACUACION

UFLA

AL FINALIZAR SU TURNO DESCONECTE LA MAQUINA

MATERIAL

IKAY C-MAG 150 A



LABORATORIO DE QUÍMICA

AGUA PASADA POR OSMÓSMO

EVACUACIÓN

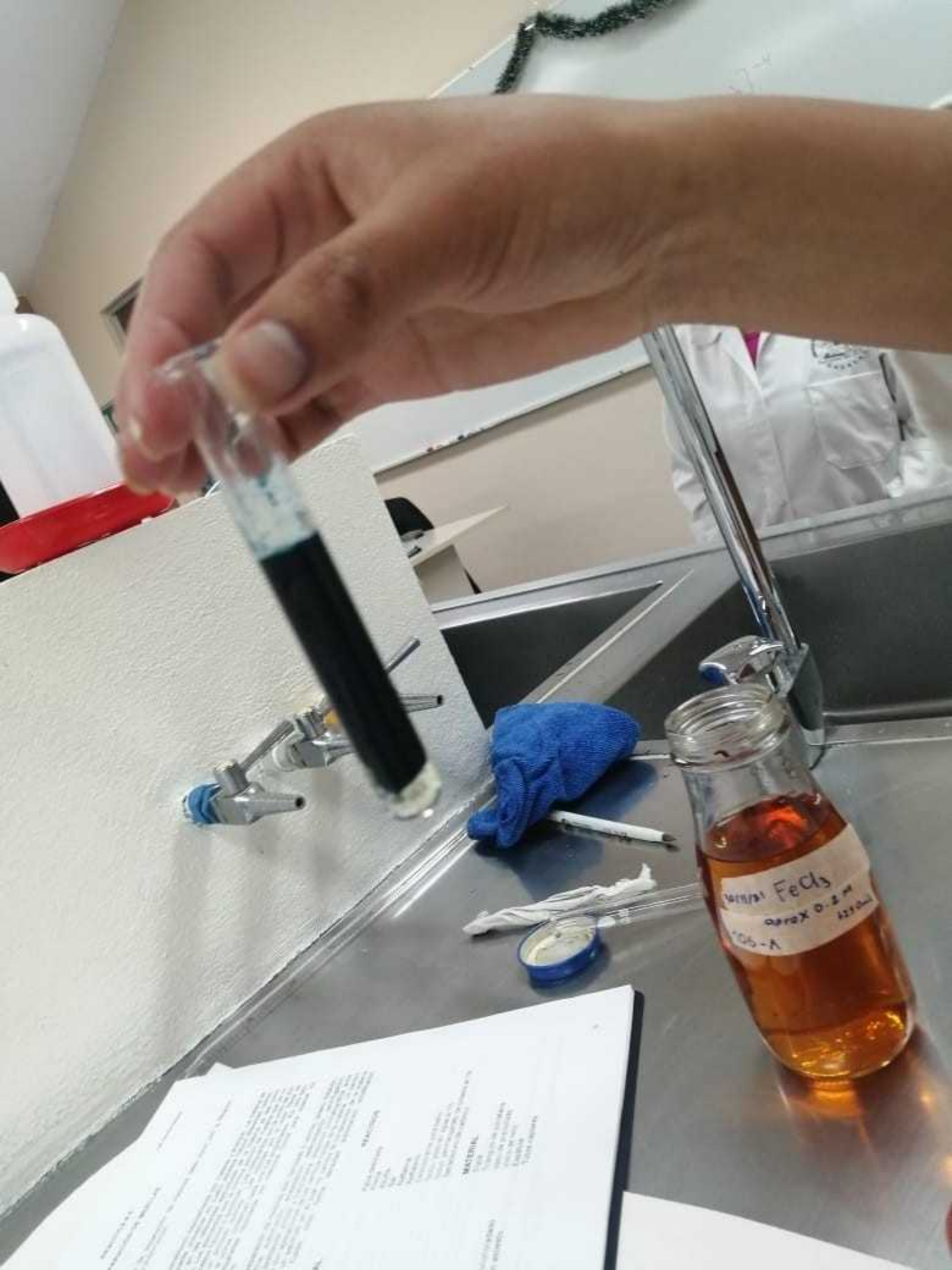
MATERIA

Reacción  
1  
Sofia P.G.

3

10mL 1/10





FeCl<sub>3</sub>  
0.2 M  
12/02

Procedure

1. Weigh out 0.2 g of FeCl<sub>3</sub> and dissolve it in 100 mL of distilled water in a 100 mL volumetric flask.

2. Add 10 mL of 1 M HCl to the solution.

3. Dilute the solution to the 100 mL mark with distilled water.

4. Store the solution in a clean, dry container.

5. Use the solution for the experiment.

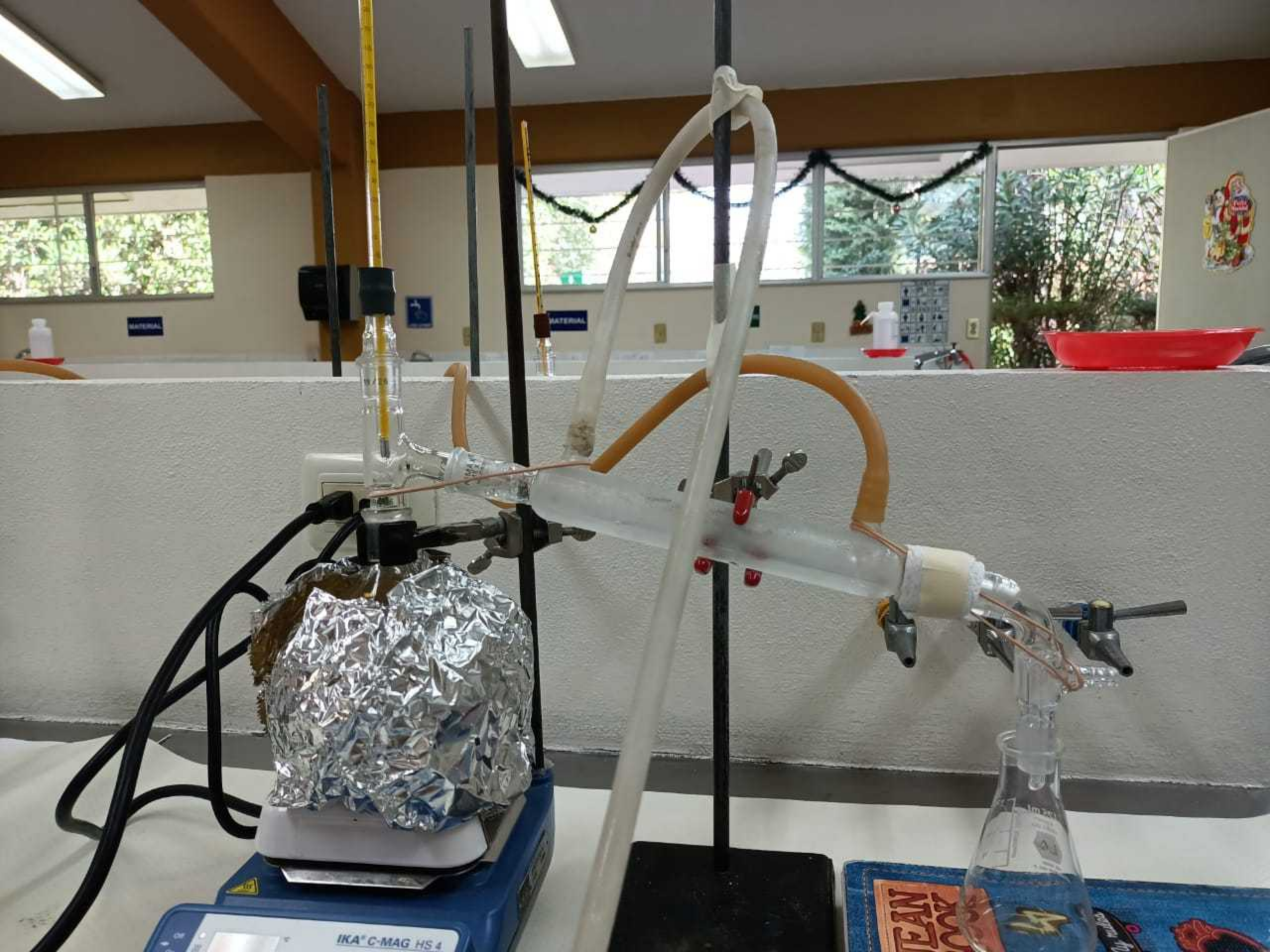
Materials

- FeCl<sub>3</sub>
- Distilled water
- 1 M HCl
- 100 mL volumetric flask
- 10 mL graduated cylinder
- 100 mL beaker
- 100 mL Erlenmeyer flask
- 100 mL graduated cylinder
- 100 mL beaker
- 100 mL Erlenmeyer flask



OFICINA





IKA® C-MAG HS 4

LEAN  
LOOK